EXPERIMENTAL

Cultivation. L. lepideus FPRL 7B (Forest Products Research Laboratory, Princes Risborough, U.K.) was cultivated on a defined synthetic medium containing glucose (2%), iso-leucine (0.15%), and mineral salts [3] for 90 days.

Identification of sesquiterpene hydrocarbons. The volatile metabolites were obtained by circulation steam distillation [8] in pentane. The hydrocarbon fraction was separated from the more polar components by modified dry-CC [4]. The identification resulted from GC/MS data by comparison with lit. data [5], and retention time comparison of R_I values, and comparison with commercial essential oils of well-known composition (e.g. copaiva balsam), respectively. Preparation of single components for IR by prep. TLC is described in ref. [3]; IR spectral data were compared with lit. data [7]. GC conditions are described in [9], using additionally capillary columns (Carbowax 20 M) of different lengths, and a computing integrator.

Acknowledgement—This work was supported by the Deutsche Forschungsgemeinschaft.

REFERENCES

- 1. Sprecher, E. (1961) Arch. Microbiol. 38, 299.
- 2. Sprecher, E. (1980) Pharm. Ztg. 125, 1008.
- 3. Hanssen, H.-P. (1979) Dissertation, Hamburg.
- 4. Kubeczka, K.-H. (1973) Chromatographia 6, 106.
- Stenhagen, E., Abrahamsson, S. and McLafferty, F. W. (1976) Registry of Mass Spectra Data. Wiley & Sons, New York.
- Laseter, J. L., Weete, J. D. and Walkinshaw, C. H. (1973) Phytochemistry 12, 387.
- Wenninger, J. A., Yates, R. L. and Dolinsky, M. (1967) JAOAC 50, 1313.
- 8. Sprecher, E. (1963) Dtsch. Apoth. Ztg. 103, 213.
- Sprecher, E., Kubeczka, K.-H. and Ratschko, M. (1975) *Arch. Pharm.* 308, 843.

Phytochemistry, Vol. 21, No. 5, pp. 1160-1162, 1982. Printed in Great Britain.

0031-9422/82/051160-03\$03.00/0 © 1982 Pergamon Press Ltd.

SESQUITERPENES AND NORSESQUITERPENES FROM PECHUEL-LOESCHEA LEIBNITZIAE*

FERDINAND BOHLMANN and NALEEN BORTHAKUR

Institute for Organic Chemistry, Technical University of Berlin, D-1000 Berlin 12, West Germany

(Received 9 September 1981)

Key Word Index—Pechuel-Loeschea leibnitziae; Compositae; Inuleae; sesquiterpenes; guaiene derivative; norsequiterpenes; guaianolide.

Abstract—The aerial parts of *Pechuel-Loeschea leibnitziae* afforded, in addition to thymohydroquinone dimethyl ether, sitosterol and stigmasterol, xerantholide and its probable precursor, the 11,13-dihydro derivative methyl pechueloate and two norsequiterpenes, probably formed by degradation of the corresponding sesquiterpene acid. Two eudesmane derivatives were also isolated. The chemotaxonomy of *Pechuel-Loeschea* is discussed briefly.

INTRODUCTION

The monotypic genus *Pechuel-Loeschea* has not been investigated chemically. This genus was included in *Pluchea* [1], but was later maintained to be a separate genus [2, 3]. It was of interest, therefore, to see whether the chemistry supports a separation from *Pluchea* or not.

RESULTS AND DISCUSSION

The roots of P.-L. leibnitziae (Kuntze) O. Hoffm. gave only thymohydroquinone dimethyl ether (1).

*Part 424 in the series "Naturally Occurring Terpene Derivatives". For Part 423 see Bohlmann, F., Jakupovic, J. and Vogel, W. (1982) *Phytochemistry* 21, 1153.

Whilst the aerial parts afforded it as the main compound, stigmasterol, sitosterol, the acid 2a[4], the corresponding methyl ester 2b and the isomeric ester 3 were also produced. The structure of compound 3 followed from the ¹H NMR data (Table 1). The

Short Reports 1161

Table 1. ¹H NMR spectral data of compounds 3, 6 and 7 (400 MHz, TMS as int. standard)

	3	6	7
H-1	3.18 ddd(br)	3.26 ddd(br)	2.96 ddd(br)
$H-2\alpha$	2.59 ddd	2.60 dd	2.41 dd
Η-2β	2.06 d(br)	2.17 dd	2.03 dd
Η-6α	2.82 dd(br)		2.78 dddd(br)
	}	$2.73 \ t(br)$	
Η-6β	2.44 dd(br)		2.25 dd(br)
Η-7α	,	2.80 dddd }	
}	2.91 dd(br)	}	1.31 m
Η-7β		2.43 dddd	
H-8		—)	
}	1.63 m	}	3.90 dddd
H-8'		_	
Η-9α		2.82 dd	1.38 ddd
	1.78 m		
Η-9β		2.52 ddd	1.53 d(br)
H-10	2.13 dddg	2.33 dddq	2.15 dddq
H-13	6.20 s	_	_
H-13'	5.62 s	_	_
H-14	0.65 d	0.71 d	1.10 d
H-15	1.63 d	1.72 dd	1.70 d
OMe	3.77 s		

J(Hz): Compound 3: $1,2\alpha = 7$; $1,2\beta \sim 1$; $1,10 \sim 3$; 1,15 = 2; $2\alpha,2\beta = 19$; $2\alpha,10 = 1$; $6\alpha,6\beta = 19$; $6\alpha,7 \sim 1.5$; $6\beta,7 = 12$; $6\alpha,8\alpha \sim 1$; $7,8\alpha \sim 1.5$; $7.8\beta = 12$; $9\alpha,10 = 9\beta,10 \sim 3$; 10,14 = 7; compound 6: $1,2\alpha = 7$; $1,2\beta = 2$; $1,10 \sim 7$; $2\alpha,2\beta = 18$; $6\alpha,6\beta = 14$; $6\beta,15 = 10,15 = 2$; 6.5; 7.7' = 13; $7\beta,9\beta = 1$; 9,9' = 13.5; $9\alpha,10 = 6.5$; $9\beta,10 = 4$; 10,14 = 7; compound 7: $1,2\alpha = 6$; $1,2\beta = 3.5$; 1,10 = 7; $1,2\beta = 10$; $1,2\beta = 10$;

O

A

$$X = CH_2$$

S

 $X = \alpha - Me$

A

 $X = CH_2$

A

 $X = CH_2$

A

 $X = \alpha - Me$

A

 $X = CH_2$

A

 $X = \alpha - Me$

B

 $X = \alpha - Me$

A

 $X = \alpha - Me$

B

 $X = \alpha - Me$

A

 $X = \alpha - Me$

B

 $X = \alpha$

presence of an ester group, already indicated by the mass spectrum, followed from the methoxyl signal, while the methylene protons (H-13) showed signals

Table 2. ¹H NMR spectral data of compounds 4, 5, 8 and 9 (400 MHz, CDCl₃ TMS as int. standard)

	4	5	8	9
H-1	3.20 dddd	3.13 dddd	3.13 <i>dddd</i>	3.26 dddd
Η-2α	2.66 ddd	2.63 ddd	2.65 ddd	2.68 ddd
Η-2β	$2.13 \ d(br)$	$2.09 \ d(br)$	$2.12 \ d(br)$	$2.11 \ d(br)$
Η-6α	$3.15 \ d(br)$	$2.98 \ d(br)$	$2.35 \ d(br)$	2.43 d(br)
Η-6β	2.49 dd(br)	2.38 m		2.35 m
		}	2.63 m	
H-7	3.05 ddddd	2.08 m		3.62 ddd
H-8	4.14 ddd	4.15 ddd	5.18 ddd	4.15 ddd
Η-9α	1.95 ddd	1.82 <i>ddd</i>	1. 94 ddd	2.05 ddd
Η-9β	2.43 m	2.38 m*	2.61 ddd	2.53 ddd
H-10	2.32 dddg	2.27 dddg	2.33 m	2.35 m
H-11		2.38 m*	_	_
H-13	6.30 d		2.31 ddd	1.91 ddd
	}	1.29 d		
H-13'	5.61 d		1.50 ddd	1.80 ddd
H-14	0.76 d	0.75 d	0.82 d	0.76 d
H-15	1.75 ddd	1.71 ddd	1. 64 <i>ddd</i>	1.65 ddd
H-16			4.91 ddd	4.87 ddd
H-16!	_		4.68 ddd	4.77 ddd

*In C_6D_6 H-11 dq 1.65 (J = 11, 7), H-9 1.96 ddd (J = 13, 3.5, 3.5).

J(Hz): Compound 5: 1, $2\alpha = 7$; 1, $2\beta \sim 1$; 1, $10 \sim 7$; 1, 15 = 6, $15 \sim 1.5$; 2α , $2\beta = 19$; 6α , $6\beta = 19$; 6α , $7 \sim 3$; 6β , 7 = 10; 7, 8 = 9.5; 7, 11 = 11; 8, $9\alpha = 12$; 8, $9\beta = 3.5$; 9α , $9\beta = 13$; 9α , 10 = 3.5; 9β , 10 = 4; 10, 14 = 11, 13 = 7; compounds 8 and 9: 1, $2\alpha \sim 7$; 1, $2\beta \sim 1$; 1, $10 \sim 7$; 2α , $2\beta = 6\alpha$, $6\beta = 19$; 6α , 7 = 3; 6β , 7 = 12; 7, 8 = 10; 8, $9\alpha = 12$; 8, $9\beta \sim 2$; 9α , $9\beta = 13$; 9α , 10 = 3.5; 9β , 10 = 4; 10, 14 = 7; 13, 13' = 13; 13, 16 = 4.5; 13, 16' = 9; 13', 16 = 9; 13', 16' = 7; 16, 16' = 17.

with the typical chemical shifts and very few allylic couplings. The position of the second oxygen function, a conjugated keto group, could also be deduced from the spectral data. The proton which displayed a broadened three-fold doublet at δ 3.18 was coupled with the olefinic methyl group, with two downfield proton signals, obviously those α - to the keto group, and with a proton which was further coupled with the secondary methyl group (0.65 d). These observations clearly indicated the presence of a guaiene derivative with a 3-keto group and a 4,5-double bond. Further spin decoupling showed that the side-chain was, as usual, equatorially orientated at C-7, while the β -orientation of the 10-methyl group followed from the couplings. Therefore, compound 3 was the precursor of xerantholide (4). Two further compounds, the diketone 6 and the keto alcohol 7, showed in part similar ¹H NMR spectral data (Table 1). Oxidation of 7 afforded 6. Accordingly, both had the same carbon skeleton and identical stereochemistry. The α -orientation of the hydroxyl group in 7 followed from the couplings, while its position was assigned by spin decoupling, which further showed that the conjugated keto group had the same position as in 3. Compounds 6 and 7 were degradation products of 3. The latter we have given the name methyl pechueloate. In addition we have named 6 norpechuelone and 7 norpechuelol.

Two sesquiterpene lactones were also isolated, although the guaianolides 4 and 5, could not be

1162

separated completely. Addition of diazomethane gave the pyrazolines 8 and 9, which were easily separated from 5. Compound 4 was the known guaianolide xerantholide [5]. Its structure and stereochemistry clearly followed from the 'H NMR spectral data of 4 and the derivatives 8 and 9 (Table 2). Furthermore 5 had already been prepared from 4, though no stereochemistry was assigned [5]. These data were very similar to previous results[5]. Spin decoupling showed that the 11-methyl group was α -orientated since $J_{7,11}$ was 11 Hz. The presence of an 8,12-translactone followed from the chemical shift of H-8 and the couplings observed and the fact that 5 was prepared from 4. The relative stereochemistry of the pyrazolines 8 and 9 followed from the chemical shifts of H-7 and H-8. In the spectrum of 8 and 9, H-8 and H-7 respectively were deshielded by an azo group. This further supported the proposed trans-stereochemistry of the lactone ring.

The compounds isolated from *Pechuel-Loeschea* species clearly support the separation from *Pluchea*. So far nine species of this genus have been investigated chemically, most of them affording thiophene acetylenes and eudesmane derivatives of the cuauthemone type [6-9]. The main compound 1, however, is also present in *Pluchea* species [9], but is a widespread compound, having been isolated from widely differing genera.

EXPERIMENTAL

The air-dried plant material, collected in February 1981 in Transvaal (voucher 81/53 deposited in the Herbarium of the Botanic Research Institute, Pretoria) was extracted with Et_2O -petrol (1:3) and the resulting extracts were separated first by CC (Si gel) and further by repeated TLC (Si gel). The roots (150 g) gave 300 mg 1 and the aerial parts (290 g) 30 mg sitosterol, 30 mg stigmasterol, 200 mg 1, 8 mg 2a, 8 mg 2b, 13 mg 3 (Et_2O -petrol, 1:3), mixture of 20 mg 4 and 8 mg 5 (Et_2O -CH₂Cl₂, 1:5, not fully separated; 5 was obtained pure after transformation of 4 to 8 and 9 by addition of CH₂N₂ in Et_2O , 2 mg 6 (Et_2O -CH₂Cl₂, 1:5) and 14 mg 7 (Et_2O -CH₂Cl₂, 1:4).

Methyl pechueloate (3). Colourless gum, IR $\nu_{\text{max}}^{\text{CCL}}$ cm⁻¹: 1715, 1640 (C=CCO₂R, O=C-C=C); MS m/z (rel. int.): 262.157 [M]⁺ (87) (C₁₆H₂₂O₃), 247 [M - Me]⁺ (100), 230 [M - MeOH]⁺ (69), 215 [230 - Me]⁺ (27), 187 [215 - CO]⁺ (25), 176 [M - H₂C=CCO₂Me]⁺ (61), 161 [176 - Me]⁺ (30);

$$[\alpha]_{2e}^{\lambda} = \frac{589}{+24} \frac{578}{+26} \frac{546}{+45} \frac{436 \text{ nm}}{+63} \text{ (CHCl}_3; c 0.4).$$

11 β,13-Dihydroxerantholide (5). Colourless gum, IR $\nu_{\rm max}^{\rm CCL_f}$ cm⁻¹: 1795 (γ-lactone), 1710, 1650 (C=CC=C); MS m/z (rel. int.): 248.141 [M]⁺ (82) (C₁₅H₁₈O₃), 233 [M - Me]⁺ (6), 220 [M - CO]⁺ (8), 161 (65), 133 (51), 91 (71), 79 (100);

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{578 \ 546 \ 436 \ \text{nm}}{+4 \ +5 \ +226} \text{ (CHCl}_3; \ c \ 0.8).$$

Pyrazoline derivative (8). IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 1790 (γ-lactone), 1710 (C=CC=O); MS m/z (rel. int.): 260.141 [M - N₂]⁺ (25) (C₁₆H₂₀O₃), 245 [260 - Me]⁺ (10), 217 [245 - CO]⁺ (5), 161 (100):

$$[\alpha]_{2\sigma}^{\lambda} = \frac{589}{+192} \frac{578}{+190} \frac{546}{+167} \frac{436}{-95}$$
 (CHCl₃; c 0.1).

Pyrazoline derivative (9). IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 1782 (γ -lactone), 1702 (C=CC=O); MS m/z (rel. int.): 288 [M]⁺ 260.141 [M - N₂]⁺ (63), 245 [260 - Me]⁺ (12), 161 (100);

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589 \quad 578 \quad 546 \quad 436 \text{ nm}}{+8 + 11 + 16 \quad +53} \text{ (CHCl}_3; c 0.1).$$

(8 and 9 were crystalline compounds, but owing to the small amounts produced no exact mp can be given).

Norpechuelone (6). Colourless oil, IR $\nu_{\text{max}}^{\text{CCI}_4}$ cm⁻¹: 1710 (C=O); 1710, 1645 (C=CC=O); MS m/z (rel. int.): 192.115 [M]⁺ (48) (C₁₂H₁₆O₂), 177 [M-Me]⁺ (14), 149 [177-CO]⁺ (10), 122 [M-C₄H₆O]⁺ (100); $[\alpha]_D = +$ 10 (CHCl₃; c 0.2).

Norpechuelol (7). Colourless oil, IR $\nu_{\text{max}}^{\text{CC4}}$ cm⁻¹: 3610 (OH), 1700, 1640 (C=CC=O); MS m/z (rel. int.): 194, 131 [M]* (8) (C₁₂H₁₈O₂), 176 [M - H₂O]* (100), 161 [176 - Me]* (39), 148 [176 - CO]* (45), 147 [176 - CHO]* (45); $[\alpha]_{\text{D}} = +14$ (CHCl₃; c 1.4). 6 mg 7 in 1 ml CH₂Cl₂ was stirred for 30 min with 10 mg pyridine dichromate. TLC afforded 3 mg 6, identical with the natural diketone.

Acknowledgements—We thank Dr. B. de Winter and Miss M. Welman, Botanic Research Institute, Pretoria, for their help during plant collection and identification of the plant material and the Deutsche Forschungsgemeinschaft for financial support.

REFERENCES

- 1. Wild, H. (1964) Kirkia 4, 45.
- Leins, P. (1971) Bot. Jahrb. Syst. Pflanzengesch. Pflanzengeogr. 91, 91.
- 3. Merxmüller, H. (1954) Mitt. Bot. München 1, 357.
- Bohlmann, F., Jakupovic, J. and Lonitz, M. (1977) Chem. Ber. 110, 301.
- Samek, Z., Holub, M., Drozdz, B., Grabarczyk, H. and Hladon, B. (1977) Collect. Czech. Chem. Commun. 42, 2441.
- Dominguez, X. A., Franco, R., Cano. G., Villarreal, R., Bapuji, M. and Bohlmann, F. (1981) Phytochemistry 20, 2297.
- Nakanishi, K., Croud, R., Miura, L., Dominguez, X. A., Zamudia, A. and Villarreal, R. (1974) J. Am. Chem. Soc. 96, 609.
- Bohlmann, F., Burkhardt, T. and Zdero, C. (1977) Naturally Occurring Acetylenes, p. 351. Academic Press, London.
- 9. Bohlmann, F. and Zdero, C. (1976) Chem. Ber. 109, 2653.